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### N,S-Substituted Dienes from Mono(arylthio)substituted- and S-, S,S-Substituted Dienes

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## N,S-SUBSTITUTED DIENES FROM MONO(ARYLTHIO)SUBSTITUTED- AND S-, S,S-SUBSTITUTED DIENES

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*Mono(thio)substituted dienes 1a–1b gave compounds 3a–c and 5d–g with piperazine and piperidine derivatives in dichloromethane. Compounds 8, 9, and 10 were obtained from the reactions of perchlorobutadiene (6) with 1,4-butanedithiol (7) in ethanol in the presence of sodium hydroxide. Compounds 12a–b, 13a–b were obtained from the reactions of perchlorobutadiene (6) with allylmercaptan ( $\text{CH}_2=\text{CH}-\text{CH}_2-\text{SH}$ ) and mercaptoethanol ( $\text{HO}-\text{CH}_2-\text{CH}_2-\text{SH}$ ).*

**Keywords:** Disulphid; dithiol; hexachlorobutadiene; mono- and disubstituted halodienes; mono(thio)substituted nitrodiene; N,S-thiosubstituted nitrodiene; piperazine; piperidine; thioether; thiol

The reactions of nitrodiene compounds with some thiols and amines are known.<sup>1–8</sup> Previously, we have obtained N,S-substituted dienes from the reaction of mono(aryl- and alkylthio) substituted dienes with some amines (primary amine, piperazine, and piperidine, etc.).<sup>9–11</sup>

The reactions of perchlorobutadiene with thiols in DMF, DMSO, and EtOH have been reported previously.<sup>12–20</sup> Thiosubstituted halodienes were obtained from the reactions of perchlorobutadiene with methane thiol in EtOH.<sup>21</sup>

The aim of this study was to synthesize new S-, S,S- and N,S-substituted 1,3-diene compounds and to determine their structures.

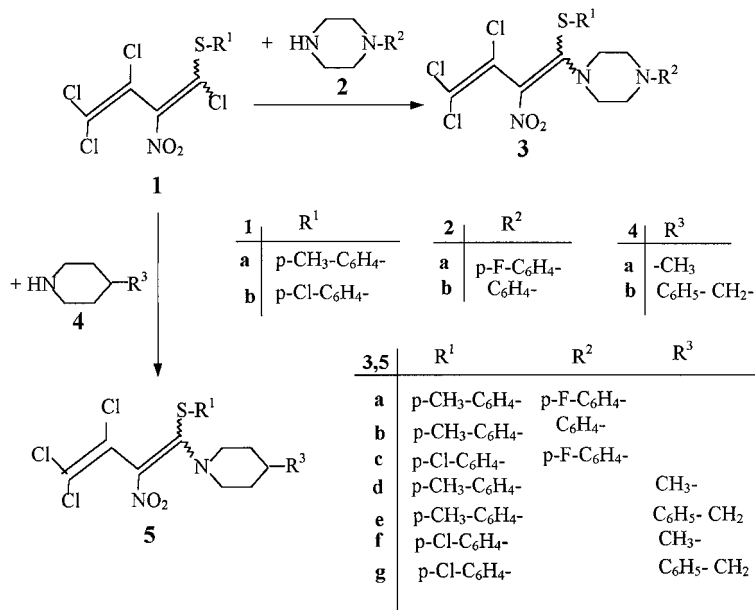
The piperazine compounds are important for therapeutical use, also some piperazine compounds were used in gen transfer reactions.<sup>22,23</sup> The piperidinyll derivatives show an excellent biological activity.<sup>24</sup>

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The occurrence of sulphur as a donor atom for transition metals is a well-known phenomenon. It acts as a very good ligating atom when in the form of the sulfide ion ( $S^{2-}$ ) or as a mercaptide ion ( $RS^-$ ), but complexes of sulphur as a thioether ( $RSR$ ) are much less abundant.<sup>25</sup>

Compounds **1a**<sup>26</sup> and **1b**<sup>27</sup> gave new compounds **3a–c** with piperazine derivatives. Compounds **5d**, **5e**, **5f**, and **5g** were obtained from the reactions of compounds **1a** and **1b** with piperidine derivatives (Scheme 1). The obtained products are yellow and stable crystals.

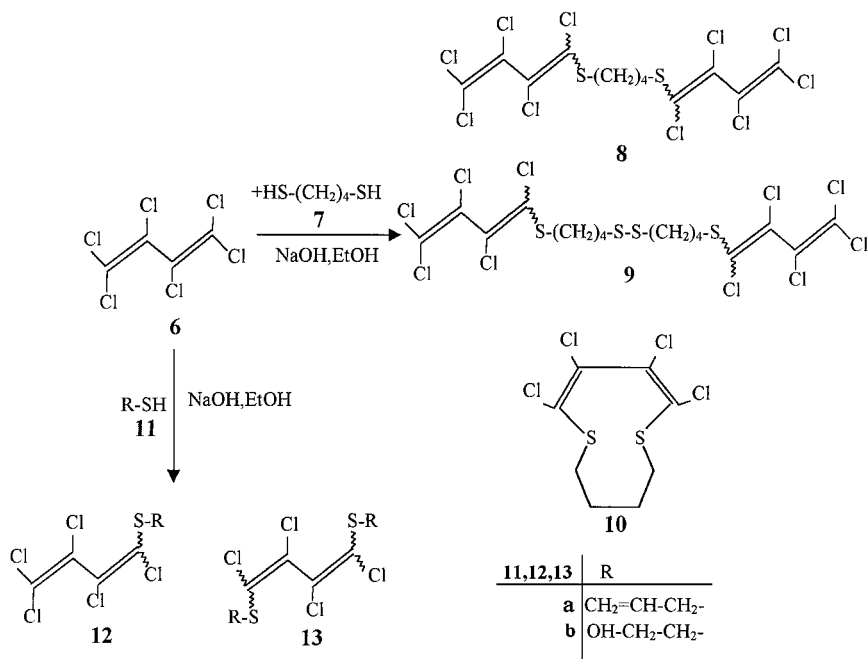


SCHEME 1

The novel compounds **8**, **9**, and **10** were obtained from the reaction of perchlorobutadiene with 1,4-butanedithiol. Compound **8** is a thioether, compound **9** is a disulphide, compound **10** is an interesting compound with a cyclic structure. The <sup>1</sup>H-NMR spectrum of compound **10** shows multiplet signals at  $\delta = 1.6$ – $2.2$  and  $2.4$ – $3.4$  ppm for the protons of the CH<sub>2</sub> groups.

Perchlorobutadiene (**6**) gave new unknown **12a–b** and **13a–b** compounds with allylthiol and 2-mercaptoethanol in EtOH in the presence of NaOH (Scheme 2).

The <sup>1</sup>H-NMR spectra of **12a–b** and **13a–b** showed the expected signals. These products are viscous oil and stable compounds. The structures of S-, S,S-, and N,S-substituted diene compounds were determined by microanalysis and spectroscopic data.



SCHEME 2

## EXPERIMENTAL SECTION

- <sup>1</sup>H-NMR: Bruker AC 200 L.
- IR: Shimadzu FTIR-8101.
- Microanalyses: Carlo-Erba 1106 Elemental Analyser.
- UV: HP 8453.
- Melting Points: Büchi SMP 20.
- Products were isolated by column chromatography on silica gel (Fluka Kieselgel 60, particle size 63–200  $\mu$ m).
- TLC plates silica 60 F<sub>254</sub> (Merck, Darmstadt), detection with ultraviolet light (254 nm).

## Preparation of S-Substituted Polyhalonitrodienes

### General Procedure I

Equimolar amounts of 1,1,3,4,4-pentachloro-2-nitro-1,3-butadiene in 10 ml of ethanol and thiols in 10 ml ethanol were mixed and NaOH (in 8 ml of water) was added at room temperature. The mixture was stirred for 24 h. Chloroform was added to the reaction mixture. The organic layer was separated and washed with water (4  $\times$  30 ml), and dried with

MgSO<sub>4</sub>. The solvent was evaporated and the residue was purified by column chromatography on silica gel.

## Preparation of N,S-Substituted Polyhalonitrodienes

### General Procedure II

Equimolar amounts of S-substituted polyhalonitrodienes and piperazine or piperidine derivatives were stirred in dichloromethane until completion of the reaction (TLC). Chloroform was added to the reaction mixture. The organic layer was separated and washed with water (4 × 30 ml), and dried with MgSO<sub>4</sub>. The solvent was evaporated and the residue was purified by column chromatography on silica gel.

*1,3,4,4-Tetrachloro-mono(p-methylphenylthio)-2-nitro-1,3-butadiene (1a)*. Compound **1a** was synthesized from 1,1,3,4,4-pentachloro-2-nitro-1,3-butadiene (2 g, 7.37 mmol) and p-methylphenylthiol (0.92 g, 7.37 mmol) according to the general procedure I. The mixture was purified by column chromatography with carbon tetrachloride as eluent.

**1a**: Yield: 2.4 g (90%); m.p. 112–113°C. UV (chloroform):  $\lambda_{\max}$  = 234, 344 nm.

*1,3,4,4-Tetrachloro-mono(p-chlorophenylthio)-2-nitro-1,3-butadiene (1b)*. Compound **1b** was synthesized from 1,1,3,4,4-pentachloro-2-nitro-1,3-butadiene (2 g, 7.37 mmol) and p-chlorophenylthiol (1 g, 7.37 mmol) according to the general procedure I. The mixture was purified by column chromatography with carbon tetrachlorid as eluent.

**1b**: Yield: 2.2 g (79%); m.p. 113–115°C. UV (chloroform):  $\lambda_{\max}$  = 245, 339 nm.

*3,4,4-Trichloro-1-(4-fluorophenylpiperazine)-1-(p-methylphenylthio)-2-nitro-1,3-butadiene (3a)*. Compound **3a** was synthesized from 1,3,4,4-tetrachloro-1-(p-methylphenylthio)-2-nitro-1,3-butadiene (0.119 g, 0.33 mmol) and 4-(p-fluorophenyl)piperazine (0.059 g, 0.33 mmol) according to the general procedure II. The mixture was purified by crystallization in methanol.

**3a**: Yield: 0.12 g (73%); m.p. 156–157°C.  $R_f$  = 0.6923 (CHCl<sub>3</sub>/Petroleumether 2:1). IR (KBr):  $\nu$  = 2980, 3010 cm<sup>-1</sup> (C–H), 1600 (C=C), 1290, 1520 (C–NO<sub>2</sub>). <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  = 6.9–7.5 ppm (m, 8H, 8 Ar–H), 2.8–3.1, 3.6–3.9 (m, 8H, 4CH<sub>2</sub>), 2.4 (m, 3H, CH<sub>3</sub>). UV (chloroform):  $\lambda_{\max}$  = 230, 388 nm. C<sub>21</sub>H<sub>19</sub>Cl<sub>3</sub>N<sub>3</sub>O<sub>2</sub>SF (502.825): calcd. C, 50.16; H, 3.80; N, 8.35; found C, 50.50; H, 3.63; N, 8.23. MS: 503.

*3,4,4-Trichloro-1-(p-methylphenylthio)-1-(N-phenylpiperazine)-2-nitro-1,3-butadiene (3b)*. Compound **3b** was synthesized from 1,3,4,4-tetrachloro-1-(p-methylphenylthio)-2-nitro-1,3-butadiene (0.12 g,

0.33 mmol) and N-phenylpiperazine (0.054 g, 0.33 mmol) according to the general procedure II. The mixture was purified by crystallization in methanol.

**3b:** Yield: 0.123 g (76%); m.p. 178–179°C.  $R_f = 0.3947$  ( $\text{CHCl}_3/\text{Petroleumether } 2:1$ ). IR (KBr):  $\nu = 2970, 3000 \text{ cm}^{-1}$  (C–H), 1600 (C=C), 1230, 1510 (C–NO<sub>2</sub>).  $^1\text{H NMR}$  ( $\text{CDCl}_3$ ):  $\delta = 6.8\text{--}7.4 \text{ ppm}$  (m, 9H, Ar–H), 2.6–3.1, 3.4–4.0 (m, 8H, 4CH<sub>2</sub>), 2.4 (m, 3H, CH<sub>3</sub>). UV (chloroform):  $\lambda_{\text{max}} = 233, 387 \text{ nm}$ .  $\text{C}_{21}\text{H}_{20}\text{Cl}_3\text{N}_3\text{O}_2\text{S}$  (484.835): calcd. C, 52.02; H, 4.15; N, 8.66; found C, 52.07; H, 3.99; N, 8.51.

*3,4,4-Trichloro-1-(p-chlorophenylthio)-1-(4-fluorophenylpiperazine)-2-nitro-1,3-butadiene (3c)*. Compound **3c** was synthesized from 1,3,4,4-tetrachloro-1-(p-chlorophenylthio)-2-nitro-1,3-butadiene (0.1 g, 0.26 mmol) and 4-(p-fluorophenyl)piperazine (0.047 g, 0.26 mmol) according to the general procedure II. The mixture was purified by crystallization in methanol.

**3c:** Yield: 0.095 g (70%); m.p. 179–180°C.  $R_f = 0.6428$  ( $\text{CHCl}_3/\text{Petroleumether } 2:1$ ). IR (KBr):  $\nu = 2980, 3010 \text{ cm}^{-1}$  (C–H), 1580 (C=C), 1280, 1500 (C–NO<sub>2</sub>).  $^1\text{H NMR}$  ( $\text{CDCl}_3$ ):  $\delta = 6.7\text{--}7.5 \text{ ppm}$  (m, 8H, 8 Ar–H), 2.8–3.2, 3.4–3.95 (m, 8H, 4CH<sub>2</sub>). UV (chloroform):  $\lambda_{\text{max}} = 251, 385 \text{ nm}$ .  $\text{C}_{20}\text{H}_{16}\text{Cl}_4\text{N}_3\text{O}_2\text{SF}$  (523.294): calcd. C, 45.91; H, 3.08; N, 8.03; found C, 45.92; H, 3.02; N, 7.93.

*3,4,4-Trichloro-1-(p-methylphenylthio)-1-(4-methylpiperidino)-2-nitro-1,3-butadiene (5d)*. Compound **5d** was synthesized from 1,3,4,4-tetrachloro-1-(p-methylphenylthio)-2-nitro-1,3-butadiene (0.2 g, 0.55 mmol) and 4-methylpiperidine (0.055 g, 0.55 mmol) according to the general procedure II. The mixture was purified by crystallization in methanol.

**5d:** Yield: 0.212 g (92%); m.p. 149–150°C.  $R_f = 0.4375$  ( $\text{CHCl}_3/\text{Petroleumether } 1:1$ ). IR (KBr):  $\nu = 2890, 2900 \text{ cm}^{-1}$  (C–H), 1600 (C=C), 1290, 1550 (C–NO<sub>2</sub>).  $^1\text{H NMR}$  ( $\text{CDCl}_3$ ):  $\delta = 7.2\text{--}7.5 \text{ ppm}$  (m, 4H, Ar–H), 3.5–4.1 (m, H, CH), 3.4–3.1 (m, 4H, 2CH<sub>2</sub>), 1.4–1.8 (m, 4H, 2CH<sub>2</sub>), 2.4 (m, 3H, S–CH<sub>3</sub>), 0.5–0.9 (m, 3H, CH<sub>3</sub>). UV (chloroform):  $\lambda_{\text{max}} = 240, 390 \text{ nm}$ .  $\text{C}_{17}\text{H}_{19}\text{Cl}_3\text{N}_2\text{O}_2\text{S}$  (421.776): calcd. C, 48.41; H, 4.54; N, 6.64; S 7.60; found C, 48.13; H, 4.20; N, 6.39; S, 7.50.

*3,4,4-Trichloro-1-(4-benzylpiperidino)-1-(p-methylphenylthio)-2-nitro-1,3-butadiene (5e)*. Compound **5e** was synthesized from 1,3,4,4-tetrachloro-1-(p-methylphenylthio)-2-nitro-1,3-butadiene (0.2 g, 0.55 mmol) and 4-benzylpiperidine (0.097 g, 0.55 mmol) according to the general procedure II. The mixture was purified by crystallization in methanol.

**5e**: Yield: 0.145 g (52%); m.p. 122–123°C.  $R_f = 0.6310$  ( $\text{CHCl}_3$ /Petroleumether 1:1). IR (KBr):  $\nu = 2970, 3000 \text{ cm}^{-1}$  (C–H), 1525 (C=C), 1240, 1510 (C–NO<sub>2</sub>).  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ):  $\delta = 6.9\text{--}7.5$  ppm (m, 9H, Ar–H), 3.2–4.1 (m, H, CH), 3.0–3.3 (m, 4H, 2CH<sub>2</sub>), 1.4–1.9 (m, 4H, 2CH<sub>2</sub>), 2.2–2.3 (m, 2H, CH<sub>2</sub>), 2.4 (m, 3H, S–CH<sub>3</sub>). UV (chloroform):  $\lambda_{\text{max}} = 231, 390$  nm.  $\text{C}_{23}\text{H}_{23}\text{Cl}_3\text{N}_2\text{O}_2\text{S}$  (497.844): calcd. C, 55.49; H, 4.65; N, 5.62; S 6.44; found C, 55.20; H, 4.35; N, 5.37; S, 6.13.

*3,4,4-Trichloro-1-(p-chlorophenylthio)-1-(4-methylpiperidino)-2-nitro-1,3-butadiene (5f)*. Compound **5f** was synthesized from 1,3,4,4-tetrachloro-1-(p-chlorophenylthio)-2-nitro-1,3-butadiene (0.2 g, 0.52 mmol) and 4-methylpiperidine (0.052 g, 0.55 mmol) according to the general procedure II. The mixture was purified by crystallization in methanol.

**5f**: Yield: 0.153 g (66%); m.p. 172–173°C.  $R_f = 0.7330$  ( $\text{CHCl}_3$ /Petroleumether 1:1). IR (KBr):  $\nu = 2970, 2985 \text{ cm}^{-1}$  (C–H), 1580, 1600 (C=C), 1290, 1545 (C–NO<sub>2</sub>).  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ):  $\delta = 7.3\text{--}7.5$  ppm (m, 4H, Ar–H), 3.5–4.1 (m, H, CH), 3.1–3.4 (m, 4H, 2CH<sub>2</sub>), 1.5–1.8 (m, 4H, 2CH<sub>2</sub>), 0.6–1.1 (m, 3H, CH<sub>3</sub>). UV (chloroform):  $\lambda_{\text{max}} = 253, 389$  nm.  $\text{C}_{16}\text{H}_{16}\text{Cl}_4\text{N}_2\text{O}_2\text{S}$  (442.194): calcd. C, 43.40; H, 3.65; N, 6.33; S 7.25; found C, 43.09; H, 3.32; N, 6.22; S, 6.91.

*3,4,4-Trichloro-1-(4-benzylpiperidino)-1-(p-chlorophenylthio)-2-nitro-1,3-butadiene (5g)*. Compound **5g** was synthesized from 1,3,4,4-tetrachloro-1-(p-chlorophenylthio)-2-nitro-1,3-butadiene (0.2 g, 0.52 mmol) and 4-benzylpiperidine (0.092 g, 0.52 mmol) according to the general procedure II. The mixture was purified by crystallization in methanol.

**5g**: Yield: 0.147 g (54%); m.p. 129–130°C.  $R_f = 0.40$  ( $\text{CHCl}_3$ /Petroleumether 1:1). IR (KBr):  $\nu = 2980, 3000 \text{ cm}^{-1}$  (C–H), 1585 (C=C), 1290, 1520 (C–NO<sub>2</sub>).  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ):  $\delta = 6.9\text{--}7.5$  ppm (m, 9H, Ar–H), 3.4–4.2 (m, H, CH), 3.0–3.3 (m, 4H, 2CH<sub>2</sub>), 1.4–2.0 (m, 4H, 2CH<sub>2</sub>), 2.3–2.5 (m, 2H, CH<sub>2</sub>). UV (chloroform):  $\lambda_{\text{max}} = 250, 389$  nm.  $\text{C}_{22}\text{H}_{20}\text{Cl}_4\text{N}_2\text{O}_2\text{S}$  (513.292): calcd. C, 50.98; H, 3.88; N, 5.40; S 6.18; found C, 50.68; H, 3.51; N, 5.10; S, 5.92.

*1,2,3,4,4-Pentachloro-1-(2-hydroxyethylthio)-1,3-butadiene (12b)*. Compound **12b** was synthesized from 1,1,2,3,4,4-hexachloro-1,3-butadiene (4 g, 15.32 mmol) and 2-mercaptoethanol (2.394 g, 30.64 mmol) according to the general procedure I. The mixture was purified by column chromatography with petroleumether and chloroform as eluent.

**12b**: Yield: 1.8 g (76%); oil.  $R_f = 0.48$  ( $\text{CHCl}_3$ ). IR (film):  $\nu = 2970, 2990 \text{ cm}^{-1}$  (C–H), 1550, 1600 (C=C), 3480 (OH).  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ):

$\delta = 3.6\text{--}3.9$  ppm (m, 2H, OH—CH<sub>2</sub>), 2.6–3.1 (m, 2H, S—CH<sub>2</sub>). C<sub>6</sub>H<sub>5</sub>Cl<sub>5</sub>SO (302.435): calcd. C, 23.82; H, 1.66; found C, 24.75; H, 1.77. MS: 302.

*1,4-Bis(2-hydroxyethylthio)-1,2,3,4-tetrachloro-1,3-butadiene (13b)*. Compound **13b** was synthesized from 1,1,2,3,4,4-hexachloro-1,3-butadiene (4 g, 15.32 mmol) and 2-mercaptoethanol (7.18 g, 91.89 mmol) according to the general procedure I. The mixture was purified by column chromatography with ethylacetate and petroleumether as eluent.

**13b**: Yield: 1.373 g (26%); oil.  $R_f = 0.3333$  (ethylacetate/petroleumether 1:1). IR (film):  $\nu = 2970, 2985\text{ cm}^{-1}$  (C—H), 1530, 1620 (C=C), 3490 (OH). C<sub>8</sub>H<sub>10</sub>Cl<sub>4</sub>S<sub>2</sub>O<sub>2</sub> (344.103): calcd. C, 27.92; H, 2.92; found C, 27.82; H, 3.01.

*1-Mono-(allylthio)-1,2,3,4,4-pentachloro-1,3-butadiene (12a)*, *1,4-bis-(allylthio)-1,2,3,4-tetrachloro-1,3-butadiene (13a)*. Compounds **12a** and **13a** were synthesized from 1,1,2,3,4,4-hexachloro-1,3-butadiene (2 g, 7.66 mmol) and allylmercaptan (0.57 g, 7.66 mmol) according to the general procedure I. The mixture was purified by column chromatography with petroleumether as eluent. Column chromatography gave the compounds **12a**, **13a** as oils.

**12a**: Yield: 1.5 g (68%); oil.  $R_f = 0.7142$  (petroleumether). IR (film):  $\nu$  2900, 3000  $\text{cm}^{-1}$  (C—H), 1550 (C=C), 655 (C—Cl). <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta = 5.7\text{--}6.1$  ppm (m, H, CH=), 5.0–5.5 (m, 2H, CH<sub>2</sub>=), 3.3–4.9 (m, 2H, CH<sub>2</sub>). C<sub>7</sub>H<sub>5</sub>Cl<sub>5</sub>S (298.447): calcd. C, 28.17; H, 1.68; S, 10.74; found C, 28.36; H, 1.39; S, 11.02.

**13a**: Yield: 0.5 g (20%); oil.  $R_f = 0.2381$  (petroleumether). IR (film):  $\nu = 2900, 3020\text{ cm}^{-1}$  (C—H), 1550, 1610 (C=C), 650 (C—Cl). <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta = 5.6\text{--}6.1$  ppm (m, 2H, 2CH=), 4.9–5.4 (m, 4H, 2CH<sub>2</sub>=), 2.4–3.6 (m, 4H, 2CH<sub>2</sub>). C<sub>10</sub>H<sub>10</sub>Cl<sub>4</sub>S<sub>2</sub> (336.132): calcd. C, 35.73; H, 2.99; S, 19.07; found C, 36.04; H, 2.99; S, 19.41.

*4,4'-Butanedithio-bis(1,1,2,3,4-pentachloro-1,3-butadiene) (8)*, *1,6,7,12-tetrathio-bis-(1,2,3,4,4-pentachloro-1,3-butadienyl)-icosan (9)*, *2,3,4,5-tetrachloro-1,6-dithia-cyclodeca-2,4-dien (10)*. Compounds **8**, **9**, and **10** were synthesized from 1,1,2,3,4,4-hexachloro-1,3-butadiene (2 g, 7.66 mmol) and butandithiol (0.935 g, 7.66 mmol) according to the general procedure I. The mixtures were purified by column chromatography with petroleumether as eluent. Column chromatography gave the compounds **8**, **9**, and **10**.

**8**: Yield: 0.6 g (14%); m.p: 77–78°C.  $R_f = 0.7659$  (petroleumether). IR (film):  $\nu = 2850, 2950\text{ cm}^{-1}$  (C—H), 1550, 1600 (C=C). <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta = 1.6\text{--}2.0$  ppm (m, 4H, CH<sub>2</sub>—CH<sub>2</sub>), 2.8–3.2 (m, 4H, 2CH<sub>2</sub>S). C<sub>12</sub>H<sub>8</sub>Cl<sub>10</sub>S<sub>2</sub> (570.856): calcd. C, 25.25; H, 1.41; S, 11.23; found C, 25.22; H, 1.19; S, 10.98.



**9:** Yield: 0.01 g (1%); oil.  $R_f = 0.5500$  (petroleumether). IR (film):  $\nu = 2860, 2980, 3000 \text{ cm}^{-1}$  (C–H), 1550, 1600 (C=C).  $^1\text{H NMR}$  ( $\text{CDCl}_3$ ):  $\delta = 1.6\text{--}2.2 \text{ ppm}$  (m, 8H,  $4\text{CH}_2$ ), 2.8–3.2 (m, 8H,  $4\text{CH}_2\text{S}$ ).  $\text{C}_{16}\text{H}_{16}\text{Cl}_{10}\text{S}_4$  (691.092): calcd. C, 27.80; H, 2.33; S, 18.56; found C, 27.53; H, 1.99; S, 18.37.

**10:** Yield: 0.05 g (2%); oil.  $R_f = 0.4500$  (petroleumether). IR (film):  $\nu = 2930, 2860 \text{ cm}^{-1}$  (C–H), 1550, 1600 (C=C).  $^1\text{H NMR}$  ( $\text{CDCl}_3$ ):  $\delta = 1.6\text{--}2.2 \text{ ppm}$  (m, 4H,  $2\text{CH}_2$ ), 2.4–3.4 (m, 4H,  $2\text{CH}_2\text{S}$ ).  $\text{C}_8\text{H}_8\text{Cl}_{14}\text{S}_2$  (310.093): calcd. C, 30.99; H, 2.60; S, 20.68; found C, 30.68; H, 2.40; S, 20.41.

## REFERENCES

- [1] Yu. A. Ol'dekop, R. V. Kaberdin, and V. I. Potkin, *Zh. Org. Khim.*, **16**, 543 (1980).
- [2] R. V. Kaberdin, V. I. Potkin, and V. P. Suboch, *Zh. Org. Khim.*, **29**, 1069 (1983).
- [3] C. İbiş and Ç. Sayıl, *Phosphorus, Sulfur, and Silicon*, **92**, 39 (1994).
- [4] Yu. A. Ol'dekop and R. V. Kaberdin, *Zh. Org. Khim.*, **12**, 2039 (1976).
- [5] C. İbiş, *Phosphorus, Sulfur, and Silicon*, **118**, 49 (1996).
- [6] Yu. A. Ol'dekop, R. V. Kaberdin, and V. I. Potkin, *Zh. Org. Khim.*, **14**, 1594 (1978).
- [7] C. İbiş and Ç. Sayıl, *Phosphorus, Sulfur, and Silicon*, **86**, 55 (1994).
- [8] Yu. A. Ol'dekop, R. V. Kaberdin, E. E. Buslouskaya, and I. A. Shingel, *Zh. Org. Khim.*, **15**(6), 1321 (1979).
- [9] C. İbiş and Z. Gökmen, *Phosphorus, Sulfur, and Silicon*, **143**, 67 (1998).
- [10] C. İbiş and G. Aydın, *Sulfur Lett.*, **23**, 67 (1999).
- [11] C. İbiş and N. Yılmaz, *Phosphorus, Sulfur, and Silicon*, **159**, 87 (2000).
- [12] C. İbiş and Ç. Sayıl, *Phosphorus, Sulfur, and Silicon*, **106**, 29 (1995).
- [13] C. İbiş and Ç. Sayıl, *Synth. Commun.*, **24**, 2797 (1994).
- [14] A. Roedig and G. Zaby, *Chem. Ber.*, 1614 (1979); 1606 (1979).
- [15] C. İbiş, *Phosphorus, Sulfur, and Silicon*, **130**, 79 (1997).
- [16] C. İbiş and Ç. Gürün, *Phosphorus, Sulfur, and Silicon*, **72**, 225 (1992).
- [17] C. İbiş and Ç. Gürün, *Phosphorus, Sulfur, and Silicon*, **83**, 119 (1993).
- [18] A. Roedig, G. Zaby, and W. Scharf, *Chem. Ber.*, **110**, 1484 (1977).
- [19] C. İbiş, *Liebigs Ann. Chem.*, 1873 (1984).
- [20] C. İbiş, *Liebigs Ann. Chem.*, 1009 (1987).
- [21] Diamond Alkali Company (Ert. H. Bluestone), U.S. Pat. No. 3021370 (13 February, 1962); *Chem. Abstr.*, **57**, 3293c (1962).
- [22] I. Soladin and T. D. Heat, *Synlett.*, **7**, 619 (1996).
- [23] S. Zhao and A. K. Miller, *Tetrahedron Lett.*, **37**(26), 4463 (1996).
- [24] V. Cecchetti and A. Fravolini, *J. Med. Chem.*, **39**, 4952 (1996).
- [25] W. Rosen and D. H. Busch, *J. Am. Chem. Soc.*, **91**, 4694 (1969).
- [26] C. İbiş and F. S. Göksel, *Phosphorus, Sulfur, and Silicon*, **97**, 165 (1994).
- [27] C. İbiş, *Bull. Soc. Chim. Belg.*, **105**, 317 (1996).